

Palladium composite membranes based on sinter metal supports for use in dehydrogenation of propane

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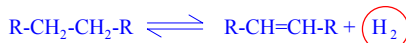
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Sponsor: BMBF

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Motivation

Thin-film composite palladium membranes can be used for hydrogen separators and dehydrogenation reactors.

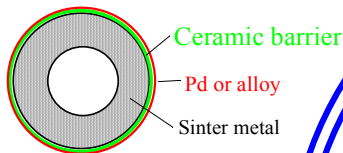


Separated by membranes in the reactor

Overcome equilibrium limitation Burning H₂ supplies heat

Advantages of composite membrane on sinter metals:

- Low palladium amount (cost)
- High H₂ permeability
- Easy module construction



Ceramic barrier:

- To avoid inter-diffusion between Pd and sinter metal.

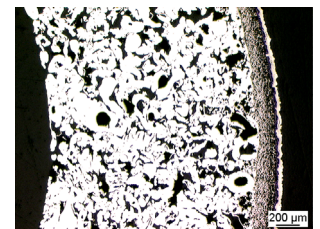
Membrane preparation

Support:

Asymmetric sinter metal tubes (coating on outer surface).

Ceramic barrier coating: Pd or Pd alloy membrane coating:

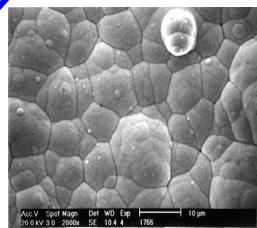
- PVD
- Wet powder spraying (WPS)
- Thermal spraying (APS: atmospheric plasma spraying)
- PVD
- Thermal spraying (APS or HVOF: high velocity oxy-fuel flame spraying)
- Electroless plating (ELP)



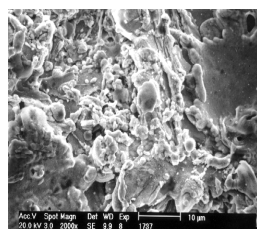
Membrane cross section

Pure Pd
Pd60%Cu40%
Pd77%Ag23% } Membranes

Pd-ELP



Pd-PVD



Surface view



Cross section view

Pd-plasma spraying

Achievements and Plans

- High H₂ permeability of ELP Pd-membranes
- Barrier properties presently limit successful plating of thinner Pd layers.

• Improvements of barrier surface roughness will allow thinner Pd membranes and thus higher H₂ fluxes.

• Membrane reactor tests under industrial process conditions are in progress, employing a new catalyst for propane dehydrogenation.

• PdCu and PdAg alloy membranes on sinter metal supports are being developed.

• Long-term tests on permeation and catalytic dehydrogenation are planned.

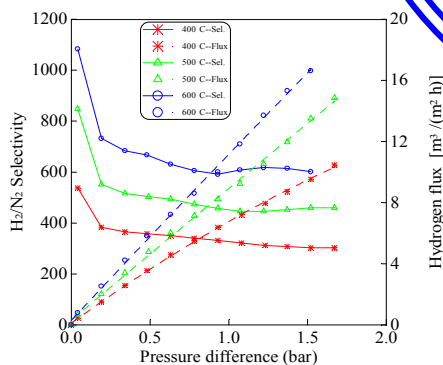
• Scale-up of membrane plating is in progress.

• Other applications of the developed membranes are being evaluated.

Typical results

Pd membrane:

- Plated by ELP
- 11-cm-long
- 1-cm diameter
- Thickness of Pd layer: ~20 μm



Permeation results in single gas test

Temperature (°C)	H ₂ Permeance F (m ³ /m ² ·h·bar ^{0.5})
400	16.6
500	22.5
600	27.9

$$J_{H_2} = F \cdot (P_{ret}^{0.5} - P_{perm}^{0.5})$$

Permeation and reaction apparatus

